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Radionuclides of natural and artificial origin

The Environmental Fernando P. Carvalho Science Education for Sustainable Human Health























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Environmental Radiation Protection

Radionuclides of natural and artificial origin

Fernando P. Carvalho



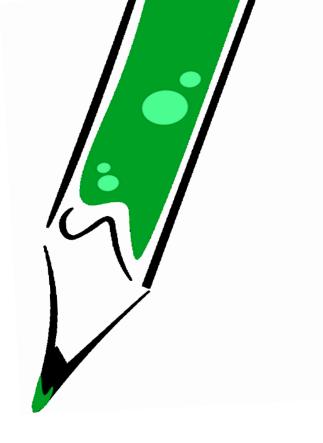
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Radioactivity and ionizing radiation

- Need for protection against ionizing radiations
- Sources of natural radioactivity

line

- Sources of artificial radioactivity
- Naturally-occuring radioactive materials (NORM) and non-nuclear industries
- Environmental monitoring for radiation protection of the population



Radioactivity and Ionizing Radiation

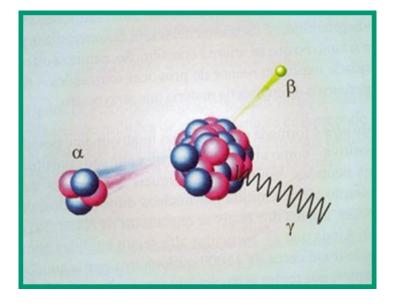
TERMINOLOGY

onuclides: Unstable nuclides

Rad

Radioactivity: Nuclear transformation of unstable nuclides with emmission of radiation (energy or particles)

Icoizing radiations: Alpha (α), Beta (β), Gamma (γ), neutrons (n), X-ray



Activity: Decay rate of a radionuclide, or disintegrations per second (1 Bq = 1 dps)

Half-life: Time to decrease to half activity of a given radionuclide

SOME HISTORICAL DATES





Marie Curie

Antoine-Henri Becquerel

- 1898 Pierry and Marie Curie isolate Radium (226Ra) and Polonium (210Po)
- 1947 E. Fermi assembles the 1st nuclear reactor (controlled chain reaction) 1947 – Alamo Gordo (New Mexico), testing of the first nuclear bomb 1956 – The 1st nuclear power plant for electricity production starts operating
- 1986 Accident at the Chernobyl nuclear power plant. 2011 - Accident at the Fukushima Nuclear Power Plan

1895 W. Roë itgen discovers X-rays

1896 **Beg**querel discovers radioactivity



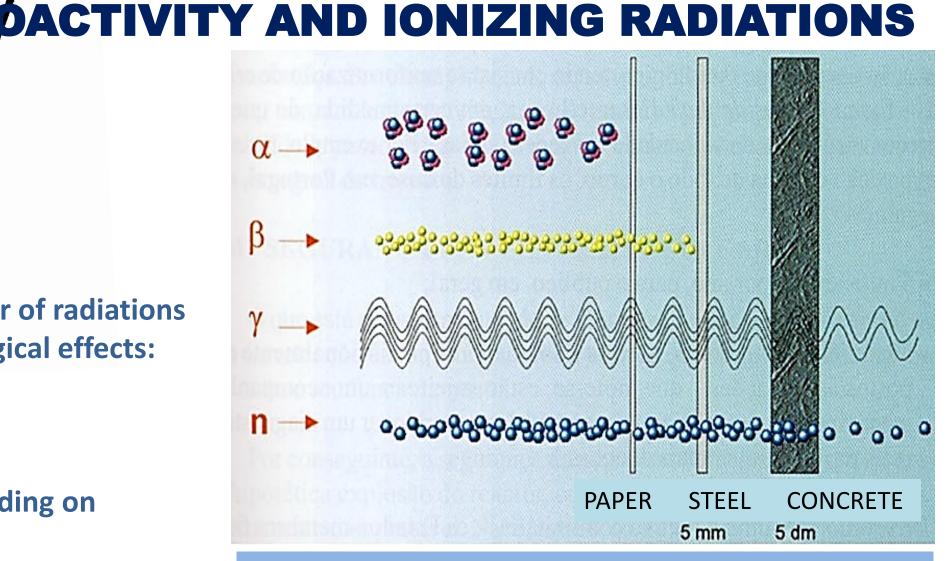
Nobel Prise 1902

Weighting factor of radiations reparding biological effects:

X-ray, γ , $\beta = 1$

n = 5-20 (depending on energy)

α = 20



Penetration of radiation in materials depends on the type of radiation and composition of materials

APPLICATIONS OF RADIATIONS AND RADIOISOTOPES

Radiography, Radiology, Radiosterilization

Rays



Radioisotopes in medicine Iodine-128, Technetium-99m, Actinium-225, Cobalt-60



Radioisotopes in industry Iridium-192, Kripton-85, Cesium-137, Selenium-75

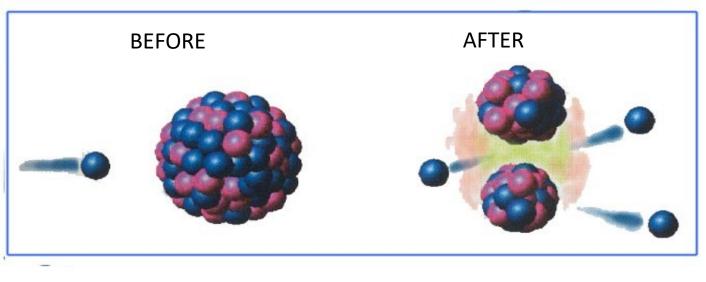


OCCUPATIONAL EXPOSURE: exposure of workers to radiation at the workplace





URANIUM ATOM FISSION

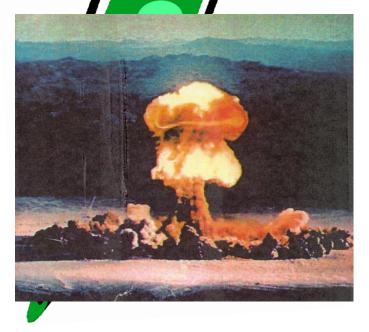


Representation of a nuclear fission reaction

• The discovery of the fission of the uranium atom started a new epoch, the atomic age

• The production of radioisotopes has started numerous applications in medicine, industry and agriculture.

PLICATIONS OF ATOMIC FISSION



Nuclear weapons: nuclear tests ¹³⁷ Cs, ⁹⁰Sr, ¹³¹I, ¹⁴⁴Ce, ... ³H, ¹⁴C

Electricity generation: nuclear power plants ⁶⁰Co, ⁶⁵Zn, ¹¹⁰Ag, ¹³⁷Cs, ^{239,240}Pu, ²⁴¹Am, ... ³H

ENVIRONMENTAL EXPOSURE:

artificial radionuclides dispersed in the environment

radiation exposure through environmental pathways

Exposure to ionizing radiation Protection against ionizing radiation



URCES OF IONIZING RADIATIONS

Cosmic radiation

Terrestrial radiation

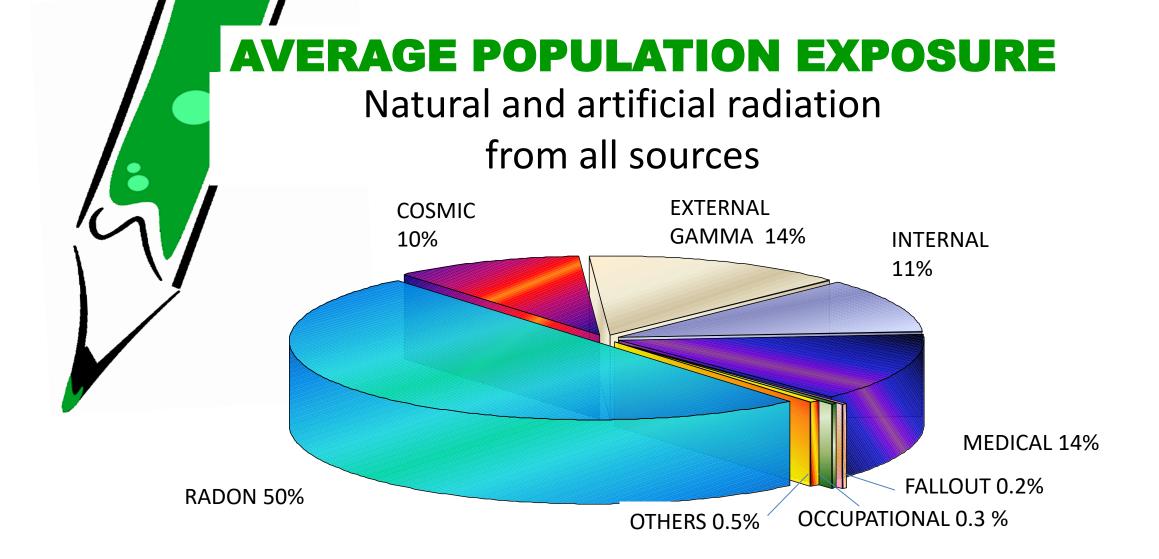
(More than 60 radioisotopes are part of the earth's crust)



Internal sources

40K, 226Ra, 210Pb, 210Po, 14C, U, Th

... Living beings formed and evolved in the presence of radioactivity



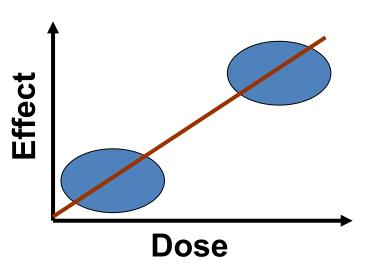
Average individual dose 2.4 mSv/a

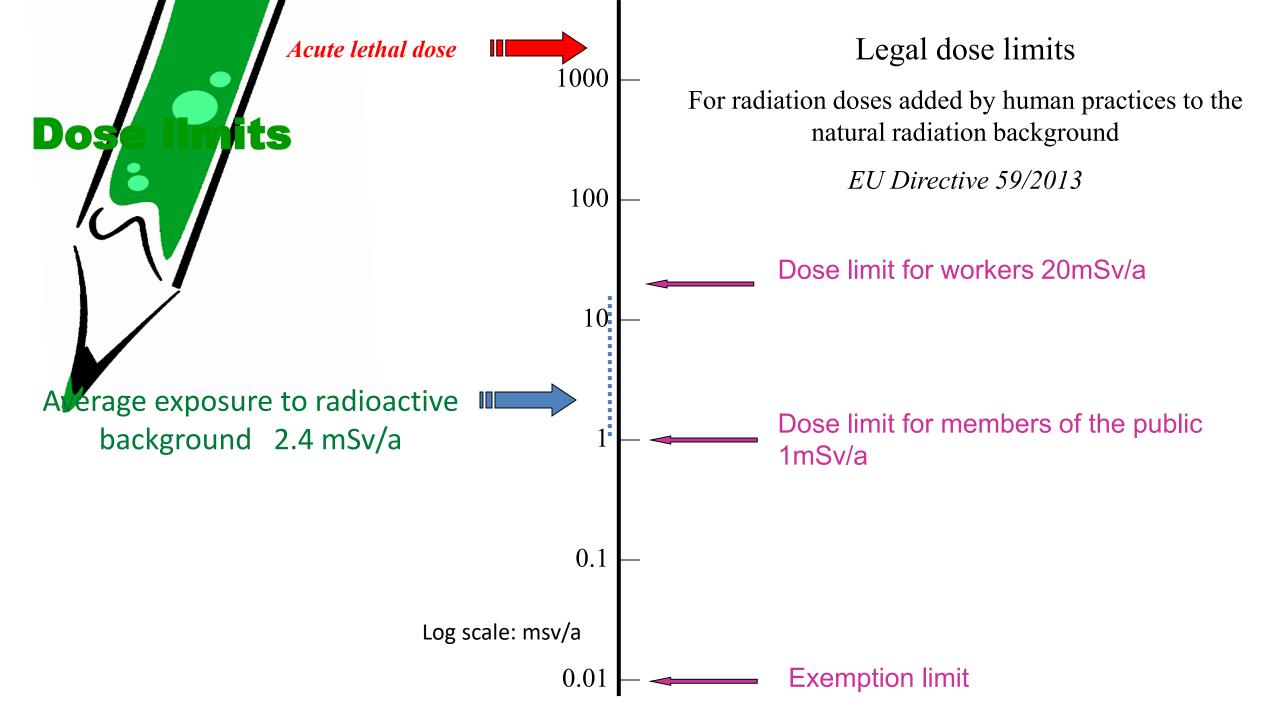
BIOLOGICAL EFFECTS OF IONIZING RADIATIONS

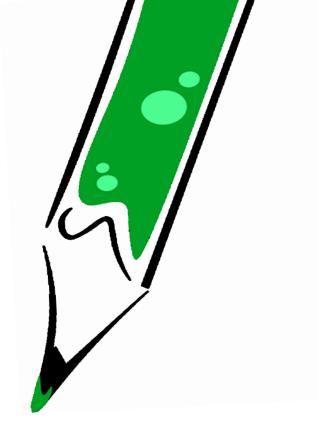
Health Effects:

- Immediate effects: exposure to high doses causes damage to human tissue and produces deterministic effects.
- Long-term effects: exposure to low doses can cause delayed cancer (latency 15-20 years) and hereditary damage. These effects are probabilistic.





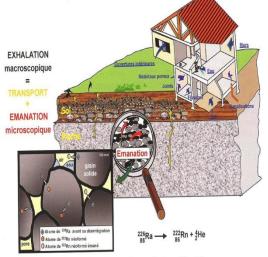




Sources of natural radionuclides and exposures

MAIN ISSUES POSED BY NATURAL RADIOACTIVITY

Radon gas indoors (homes, workplaces)

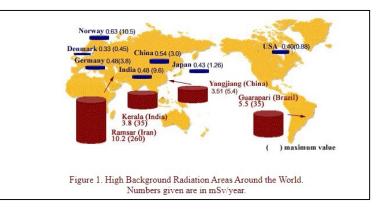


Radon formation and sources and pathways of radon from soils to the air. including indoor air

Radon is the 2nd cause of lung cancer. EU Recommended limit 300 Bq/m3 Radioactivity in drinking water (groundwater)

Regions with high natural radioactive background





WHO, EU screening limits: Total beta 1 Bq/L Total alpha 0.5 Bq/L External radiation doses may reach 40 to 60 mSv/a. Biological effects observed in populations. Not regulated yet.

Sources of artificial radionuclides and exposures

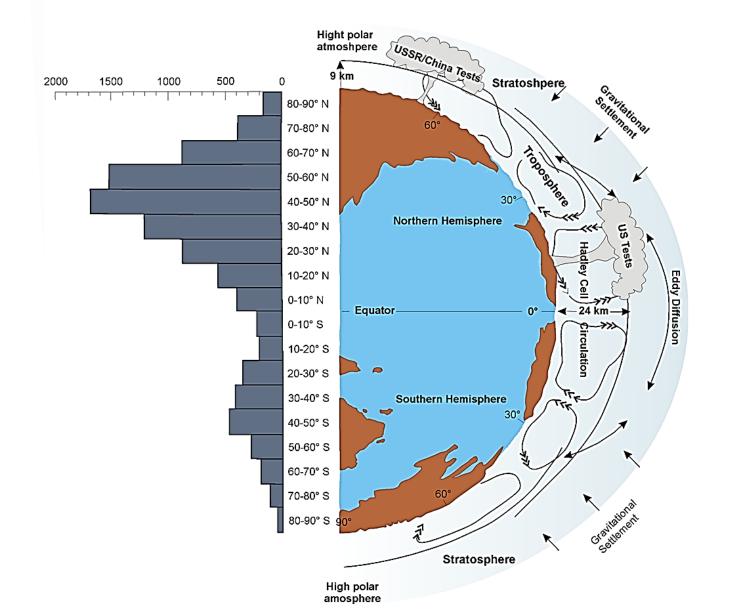
RADIOACTIVITY FROM ARTIFICIAL ORIGIN

Accidents with transport of nuclear materials Radioactive waste immersion into the seas Routine discharge of radioactive effluents into rivers and coastal seas (hospitals, industries, nuclear power plants)

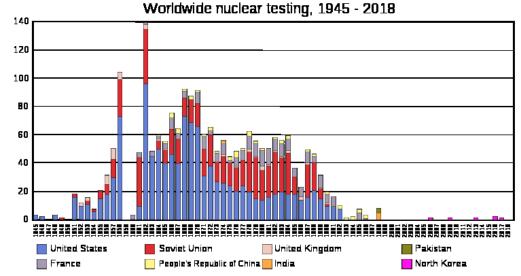
- Sinking of nuclear submarines and losses of nuclear warheads
- Nuclear accidents

NUCLEAR TESTS AND ATMOSPHERIC CIRCULATION

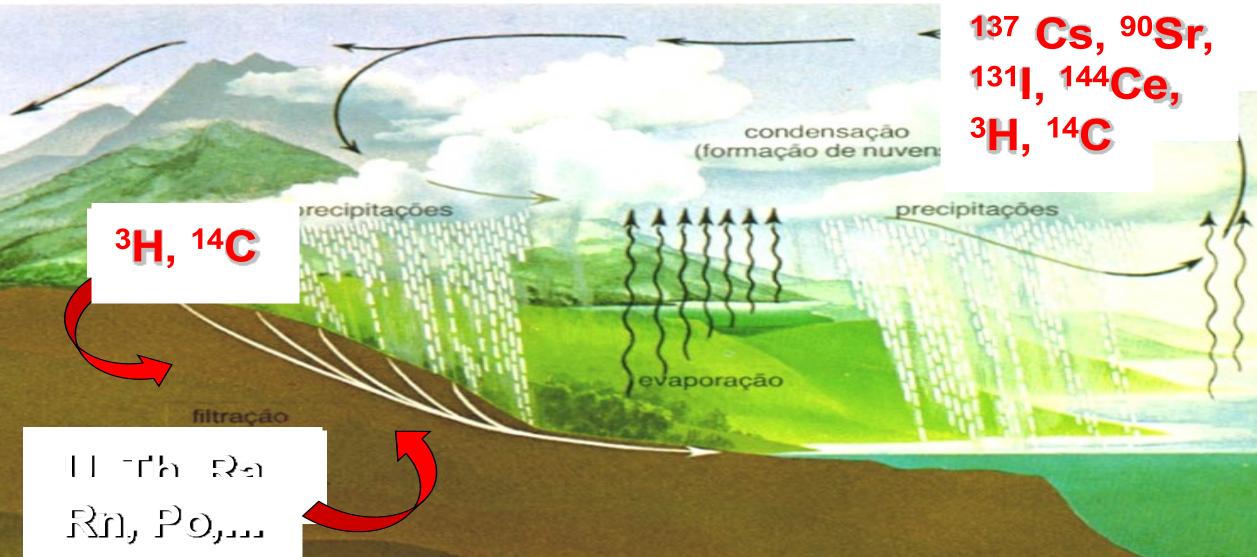
⁰Sr fallout deposit (Bq m⁻²)



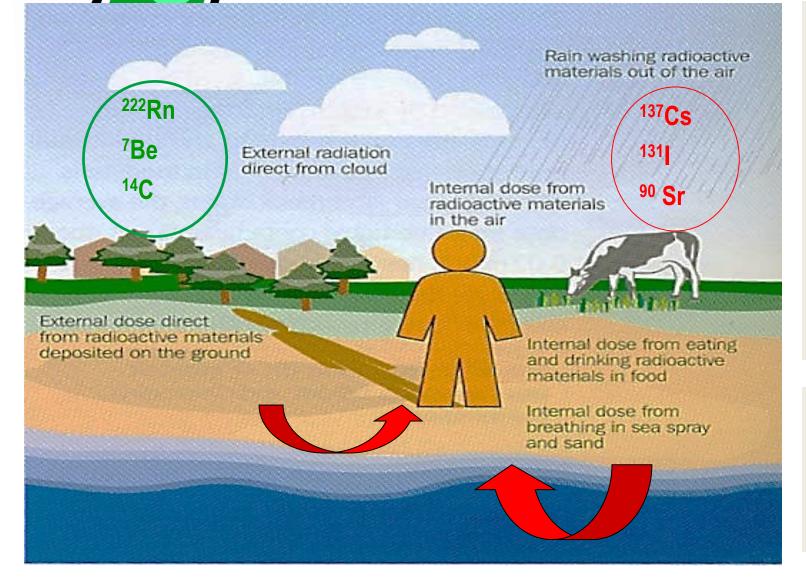
Radionuclides released into the atmosphere are transported by the atmospheric circulation and reach far regions



Water cycle and radionuclides



FOOD CHAINS AND RADIONUCLIDES TRANSFER

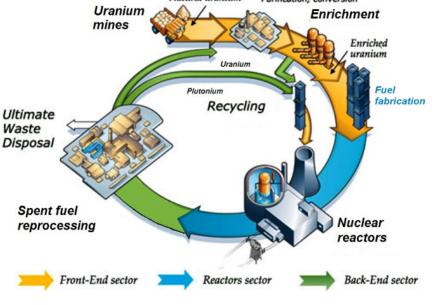


Humans may be exposed to radionuclides through:

- direct inhalation from the air
- ingestion with food and water
- external irradiation by radionuclides in clouds and deposited on the ground

Radionuclides dispersed in the environment may be re concentrated by biota and transfered in the food chains.

ARGES FROM NUCLEAR FACILITIES Nuclear Industry: Natural uranium Purification, conversion



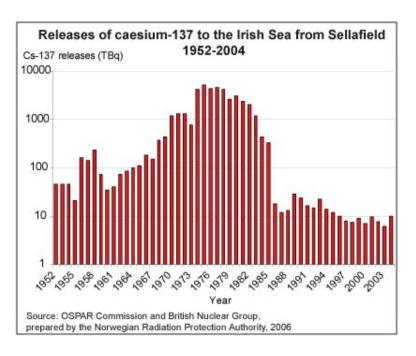
The nuclear fuel cycle



Reprocessing plants



¹³⁷Cs input to the ocean: 40 PBq



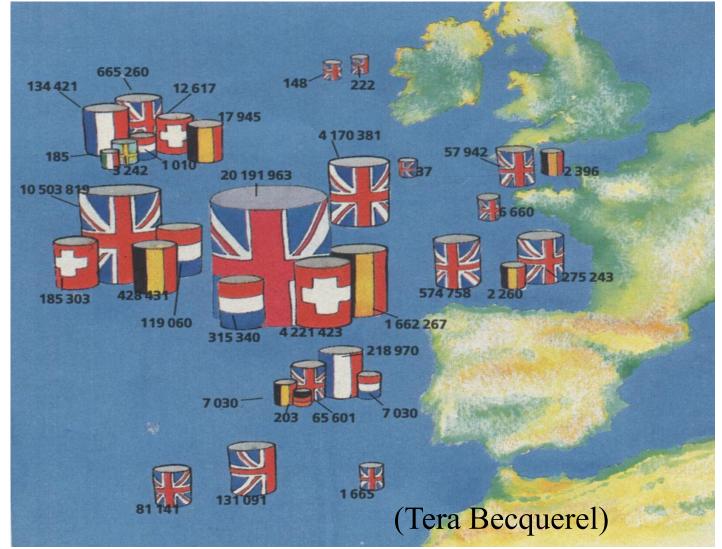
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Recipactive waste dumped into the North East Atlantic

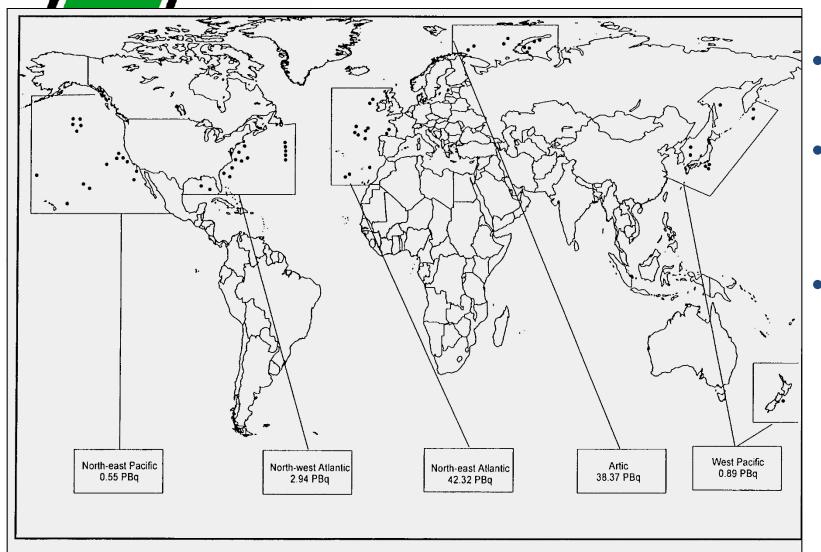
(1946-1982)



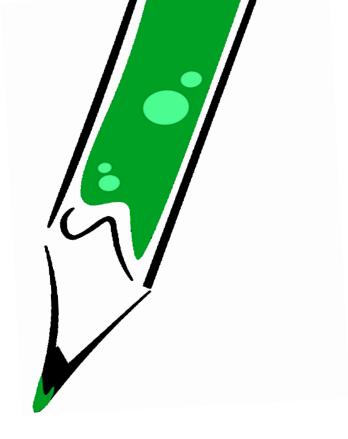
Radiactive waste from nuclear industries: Ra-226 H-3 C-14 Pu-239,240 Sr-90



Sites of immersion of radioactive waste in the ocean

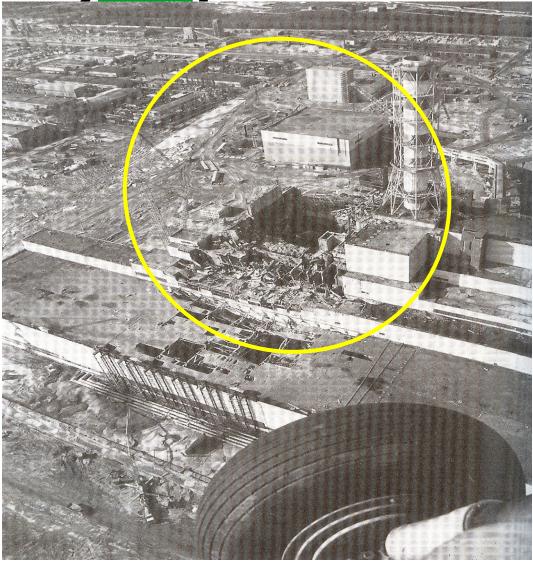


- Practice not acceptable today
- Legacy from past poor
 waste management
 policies
- Waste dumping into the sea is forbidden since 1982 (London Sea Dumping Convention)



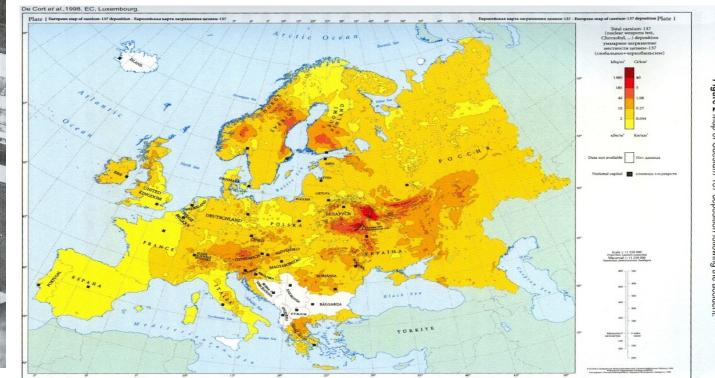
Nuclear accidents and cross-border contamination

Nuclear accident at Chernobyl

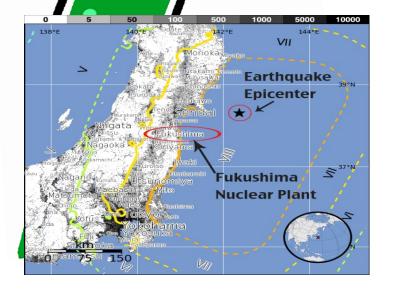


26 April 1986

Ukraine







Fukushima Daiichi NPP:

- 6 units (BWR reactors)
- Units 1-3 were in operation; units 4-6 were stopped in maintenance)
- Tsunami flooded buildings and cut power. Units 1 to 3 left with battery power only.

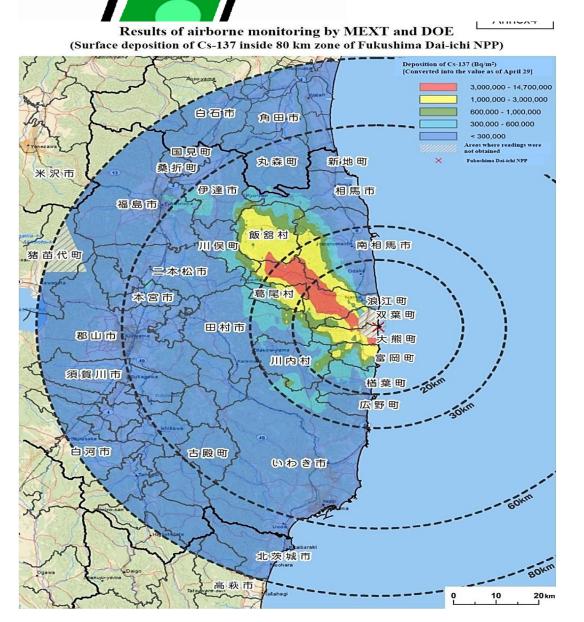
Nuclear disaster



Fukushima-Daiichi Nuclear Power Plant Information

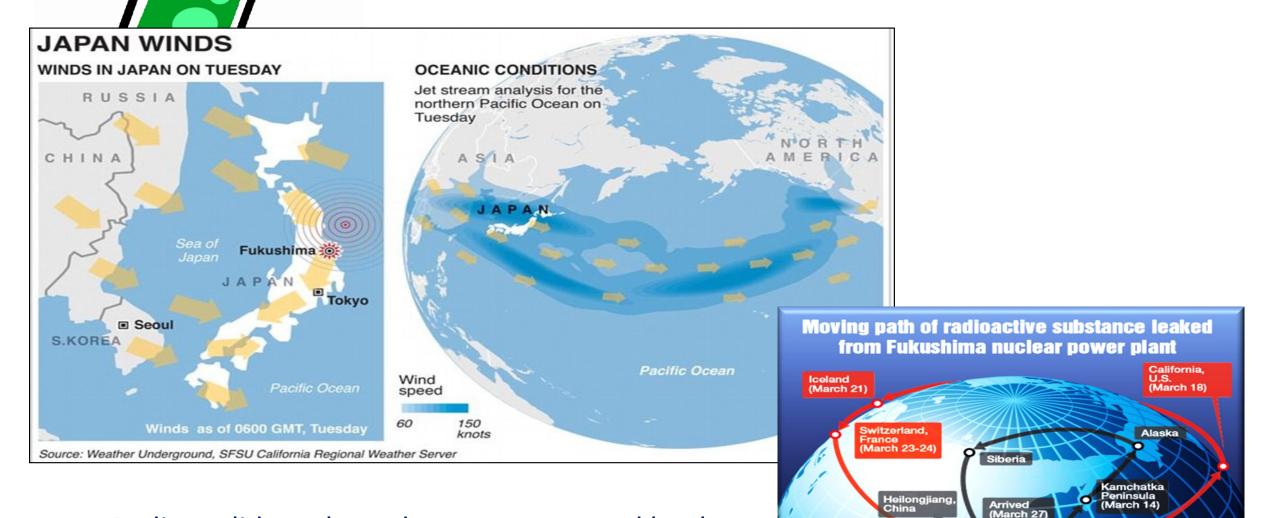
Source: Tokyo Electric Power Company

Cesium-137 deposition



- 137Cs depositions
- Soil, houses and food contaminated
- People moved to other regions
- Discharge of large amounts of contaminated water into the sea

Arborne contamination outside Japan



Expected to arrive (March 31-April 1)

KOREA

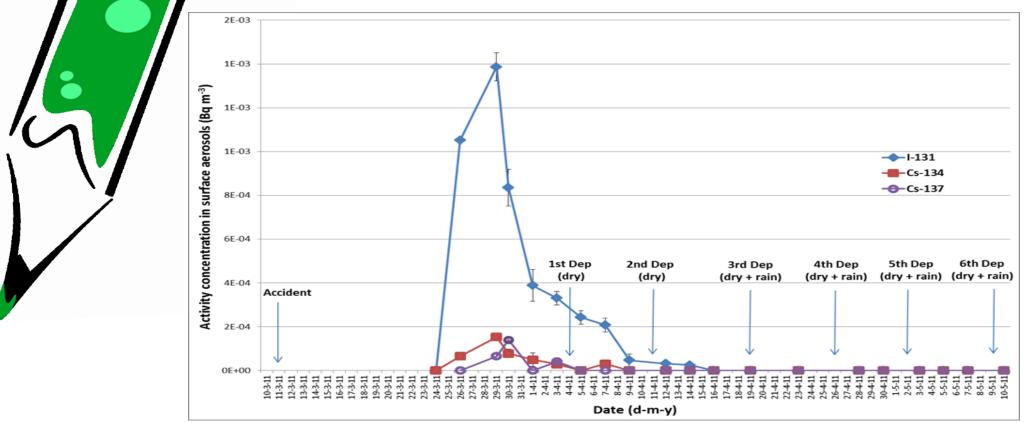
Fukushima, Japan

Source: KINS

(March 12)

Radionuclides released were transported by the global atmospheric circulation

Radionuclides measured at Lisbon



Radionuclides determined in air filters at Sacavém, Lisbon, following the Fukushima Daiichi nuclear accident. Arrows indicate the collection date of atmospheric deposition samples (Carvalho et al., J. Environ. Radioactivity 2012)



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ARTICLE

Tracking of Airborne Radionuclides from the Damaged Fukushima Dai-Ichi Nuclear Reactors by European Networks

O. Masson,^{†,*} A. Baeza,^{η} J. Bieringer,^{\perp} K. Brudecki,[#] S. Bucci,^{θ} M. Cappai,^{θ} F.P. Carvalho,^{\neq} O. Connan,[§]

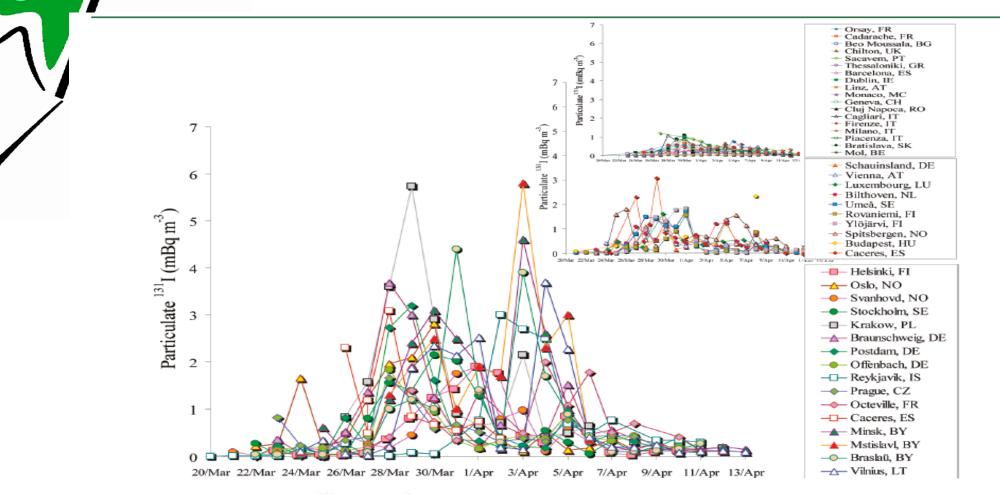
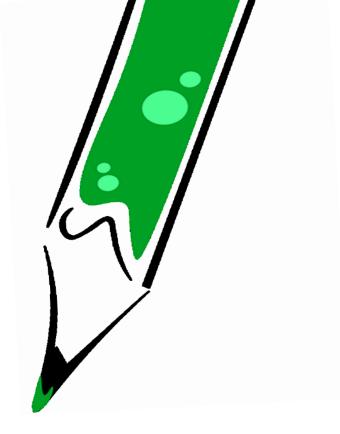


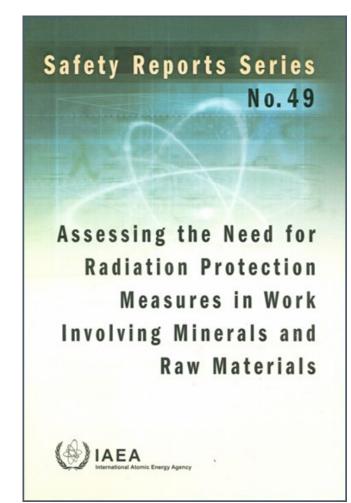
Figure 1. Time series of particulate 131 I (mBq m⁻³) in northern and central Europe (bottom), western and southern Europe (middle and top) due to the Fukushima releases.

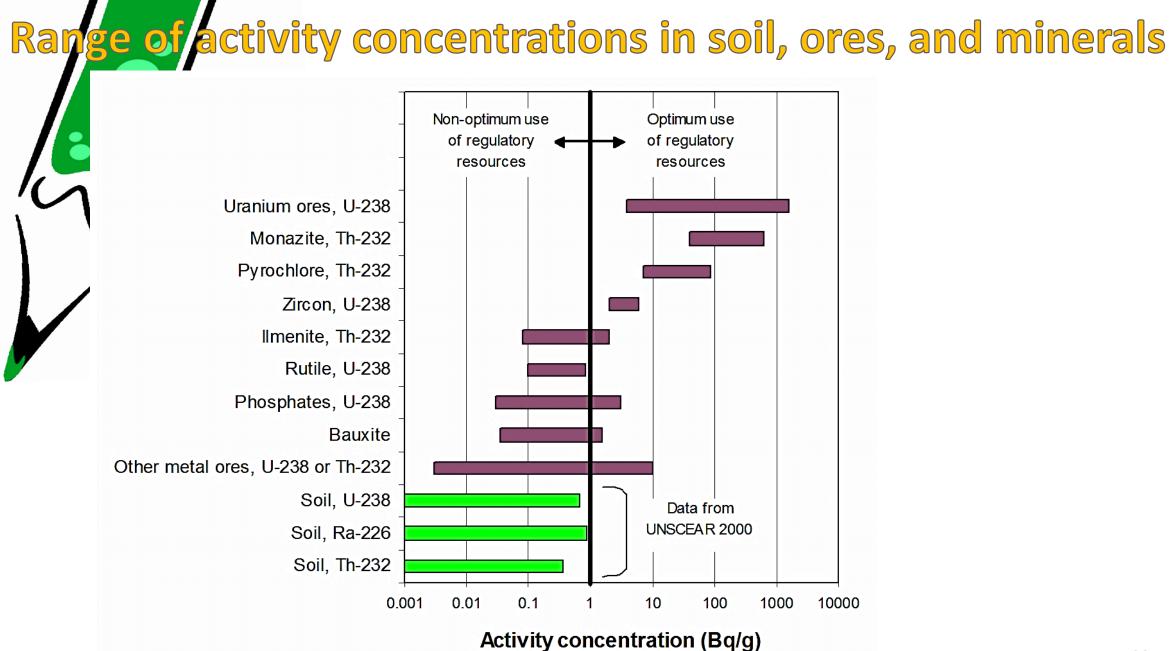


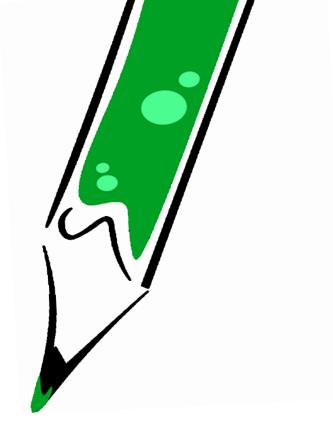
Naturally-occuring radioactive materials (NORM) and industries

dustries that process NORM

- Industries process raw materials that contain batural radionuclides
- Radionuclides may become strongly concentrated in products, wastes, or in the facilities
- Workers may be exposed to significant radiation doses
- Wastes may contain high levels of radionuclides and expose the public to radiation doses above safety limits.
- Therefore, also non-nuclear industries must be assessed to evaluate radiation risks







Environmental monitoring for radiation protection of the population

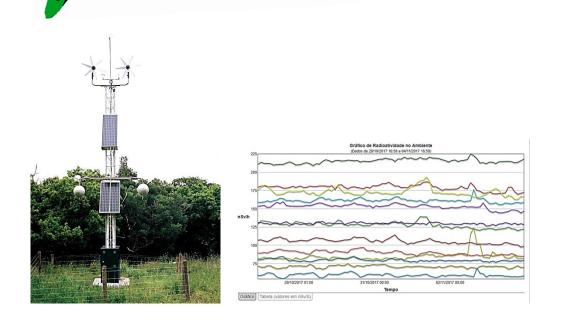
Environmental radioactivity monitoring

• Automatic measurements, by stand alone equipment

- Ambient dose rate
- Total all ha/beta counting
- Gamma spectrometry of aerosols

Advantages.

- Real time monitoring for alarm and emergency response
- International networks allow for synoptic observation



Discontinuous

- Based on sample collection and laboratory analysis
- Analysis of water, foods, soils etc.
- Accurate results for activity concentrations

Advantages:

- Follow up of the status of the environment
- Verification of compliance by industries
- Data on produts support international trade



CONCLUSIONS

The occurrence of radionuclides of natural and artificial origin requires continuous/frequent monitoring of radioactivity in foods, drinking water, and in the environment.

There are maximum limits for radioactivity in water and foods that must not be exceeded in order to protect public health.

There are dose limits for exposures from all pathways and all radionuclides, for workers and for members of the public, that are legally enforced to ensure radiation safety.



Thank you for your kind attention!

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